

Catalyst for nitrous oxide (N₂O) decomposition

CSIC has developed a novel catalyst, with a simple preparation method, that achieves the removal of 98% of N₂O emissions at temperatures below 350°C, when applied in the so-called "tail gas" of nitric acid production plants (tertiary treatment) and under real manufacturing conditions (presence of water, oxygen and at high spatial speed).

Industrial partners, from the chemical sector and interested in the production or the use of this catalyst, are being sought to collaborate through a patent licence agreement.

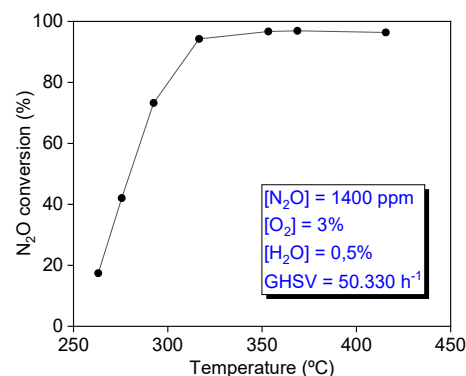
An offer for Patent Licensing

N₂O abatement at low temperatures

Nitrous oxide is a greenhouse gas with an effect 298 times greater than CO₂ on the global warming. The industrial sector generates 5% of the total emissions of N₂O and they are increasing in the last years. In particular, emissions in nitric acid plants have a particular difficulty in being reduced. The most effective technologies for the elimination of nitrous oxide in nitric acid plants are catalytic decomposition processes after the ammonia oxidation stage (secondary treatment) or in the tail gas stream (tertiary treatment).

The invention consists on a new solid catalyst for the tertiary treatment of nitrous oxide emissions. The material shows a significant improvement in the catalytic activity, allowing the use of lower volume of catalyst at lower temperatures, for the N₂O removal.

The catalyst was tested under real operating conditions of a nitric acid production plant: 1400 ppm N₂O, 3% O₂, 0.5% H₂O and a catalyst flow/volume ratio of 50,300 h⁻¹, resulting in N₂O conversion value of 98% at 340°C.



Behavior of the novel catalyst with the temperature of reaction

Main innovations and advantages

- Catalysts for tertiary treatment offer the great advantage of not influencing the heart of the nitric acid plant
- The catalyst is active at relatively low temperatures (between 250°C and 500°C)
- The catalyst can work below 300°C with conversions around 80%.
- The catalyst works with high efficiency under real conditions (in the presence of other tail gas components: O₂, H₂O)
- The catalyst has good stability: 65 h in reaction without decreasing conversion of N₂O, with the presence of H₂O and O₂.

Patent Status

Priority patent application filed suitable for international extension

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